Slide Presentation

Mass Integration for Process Design

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Mass Integration

• Tool: Minimum Mass Utility Cost for Mass Exchanger Networks with Fixed or Variable, Single Component Targets

• Objective: $\sum_i c_i L_i$

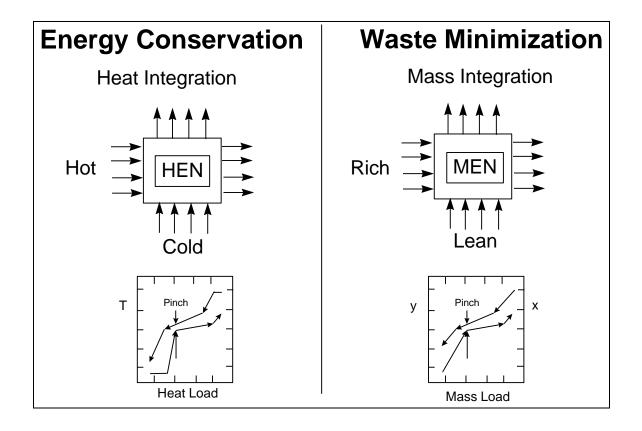
• Unit Operations: Mass Exchangers

• Framework: Conservation of Mass

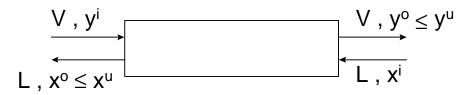
1st and 2nd Laws of Thermodynamics

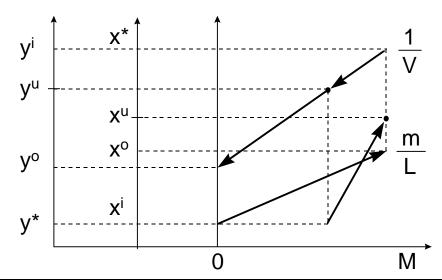
Mass cascades from high to low chemical potential

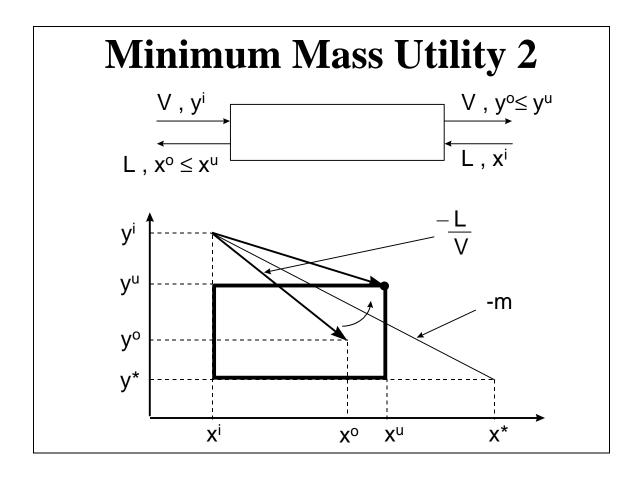
• Concepts: Mass Pinch Analysis, Composition Interval Diagrams, Mass Exchange Diagram

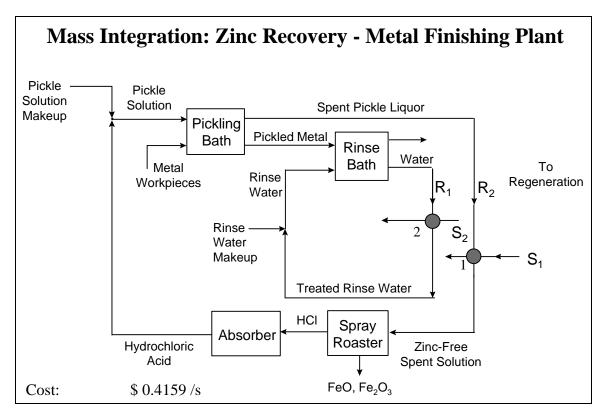


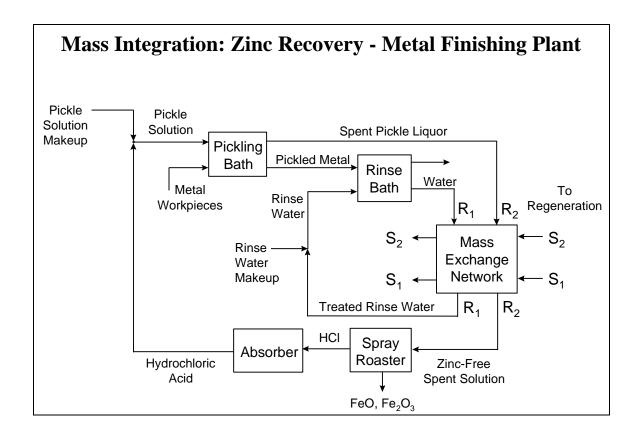
Minimum Mass Utility 1



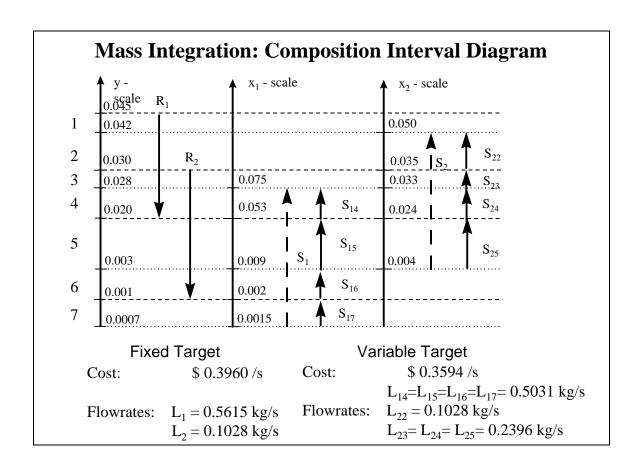


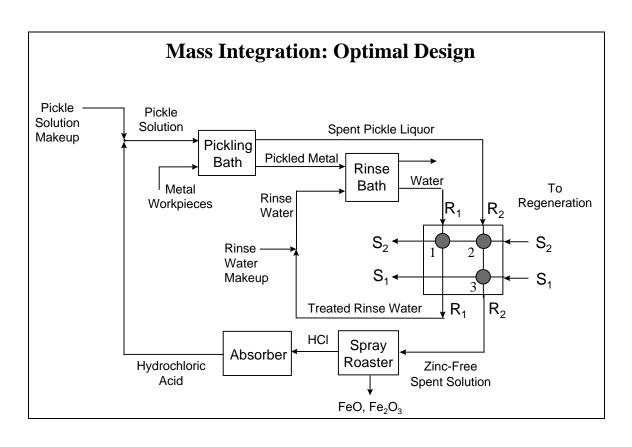






Mass Integration: System Data								
Equilibrium Data								
	Zinc Chloride							
R (water) - S_1 (Resin) $y = 0.376 (x+\epsilon) + 0.0001$, $\epsilon = 10^{-4}$ R (water) - S_2 (Phosphate) $y = 0.845 (x+\epsilon)$, $\epsilon = 10^{-4}$								
Input Output Data								
	Rich Streams Lean Streams							
V y ⁱ y ^u x ⁱ x ^u c kg/s kg/kg kg/kg kg/kg \$/kg								
R_1 R_2	0.1 1.5	0.045 0.03	0.02 0.001	S_1 S_2	0.0015 0.004	0.075 0.05	0.7 0.03	





Multicomponent Mass Integration

• Tool: Minimum Mass Utility Cost for Mass Exchanger Networks with Multicomponent Targets

• **Objective:** $\sum_{i} c_{i} L_{i}$

• Unit Operations: Mass Exchangers

• Framework: 1st and 2nd Laws of Thermodynamics

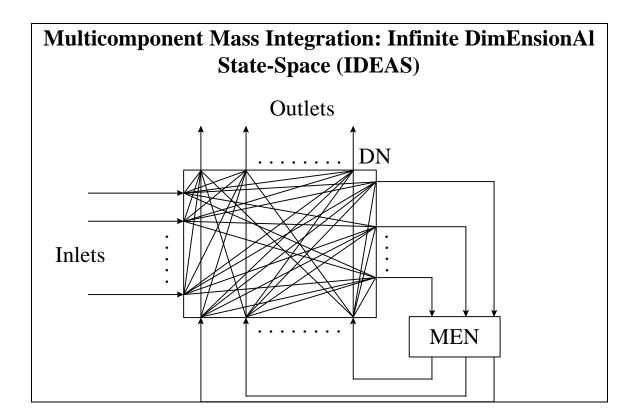
Infinite DimEnsional State Space (IDEAS)

Conservation of Mass

Mass cascades from high to low chemical

potential for each component

• Concepts: Composition Interval Diagrams, Mass Exchange Diagrams for Each Component



Multicomponent Mass Integration: Example

Input-Output Data

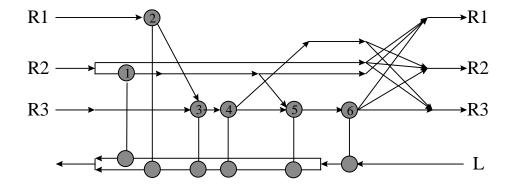
Stream	Flow	\mathbf{y}_1^{i}	$\mathbf{y}_1^{\mathrm{o}}$	$\mathbf{y}_2{}^{\mathrm{i}}$	$\mathbf{y}_2^{\mathrm{o}}$
R 1	0.125	0.040	0.016	0.042	0.002
R2	0.375	0.024	0.016	0.042	0.010
R3	0.125	0.024	0.016	0.074	0.010
	Cost	$\mathbf{X_1}^{\mathrm{i}}$	$X_1^{\text{ o}}$	$\mathbf{X}_2{}^{\mathrm{i}}$	X_2^{o}
L	1.0	0.00	0.008	0.000	0.025

Equilibrium Data

$$y_1 = 4.0 x_1$$
, $y_2 = 2.0 x_2$

Multicomponent Mass Integration: Network Design

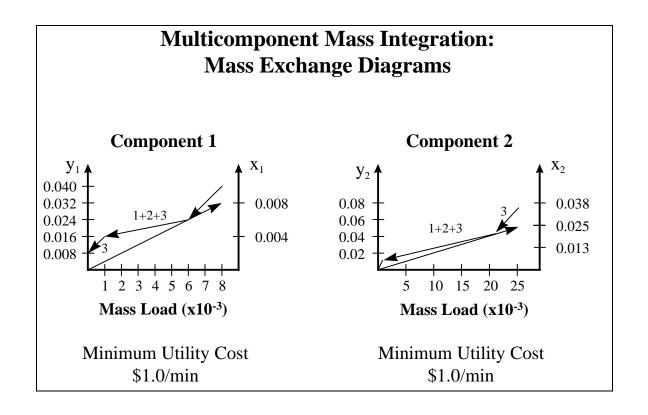
• Our recent results can identify minimum utility cost for the multicomponent MEN problem

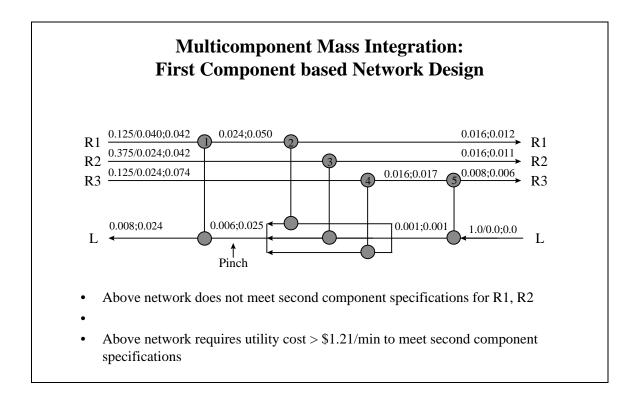


• Multicomponent Minimum Utility Cost \$1.0/min

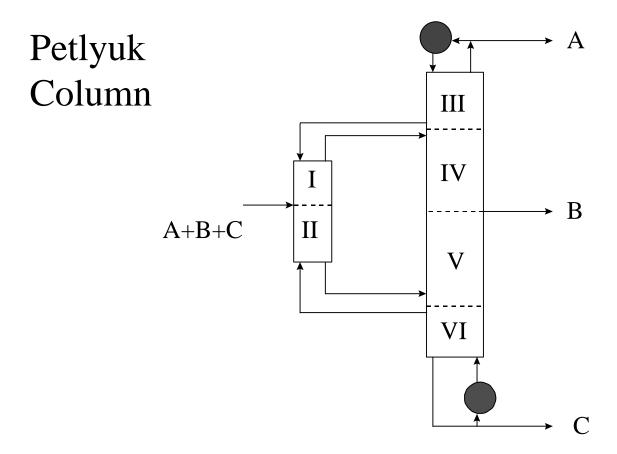
Multicomponent Mass Integration Network Data

MEX	y_1^{i}	y_1^{o}	y_2^i	$y_2^{\rm o}$	x_1^{i}	x_1^{o}	x_2^{i}	x_2^{o}	M_1	M_2
1	0.024	0.024	0.042	0.002	0.006	0.006	0.000	0.021	0.000	0.0105
2	0.040	0.024	0.042	0.074	0.006	0.010	0.037	0.029	0.002	-0.0040
3	0.024	0.024	0.074	0.010	0.006	0.006	0.005	0.037	0.000	0.0160
4	0.024	0.024	0.010	0.002	0.006	0.006	0.001	0.005	0.000	0.0020
5	0.024	0.024	0.002	0.000	0.006	0.006	0.000	0.001	0.000	0.0005
6	0.024	0.000	0.000	0.000	0.000	0.006	0.000	0.000	0.006	0.0000

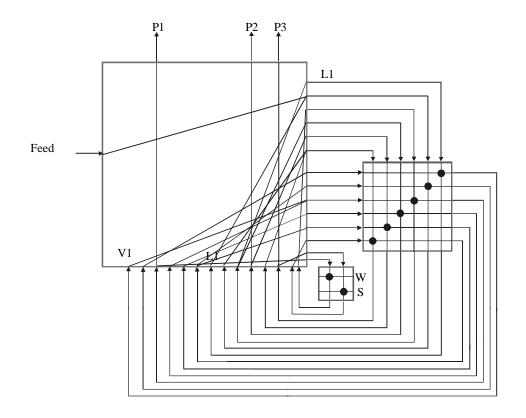




Globally Optimal Distillation Networks Minimum Utility



IDEAS Representation of Petlyuk Column



Employed assumptions:

Process is isobaric

MEX's incorporate equilibrium plates

HEX's do not result in mixed phase streams

Example:

Feed: 10 kg-moles/sec: 70 mole % N_2 , 30% O_2 .

Distillate: 7.5 kg-moles/sec: 90% N₂

Bottoms: 2.5kg-moles/sec: 90% O₂

Constant relative Volatility: 4.173

Utility cost:

Hot utility: \$105/MJ at 92°K

Cold utility: \$1030/MJ at 76°K

Pressure: 1.013x10⁵ Pa

Two cases

Case 1: Feed and product streams at 78°K (subcooled)

Case 2: Feed and distillate product are saturated vapor, and bottoms product is saturated liquid

Objective

Determine the *globally*minimum utility cost
for a given separation using IDEAS
and compare with a McCabe-Thiele
design

Case 1

Distillation Efficiency, 0_d

$$0_{d} = \frac{|-W_{min}|}{|W_{proc}|}$$

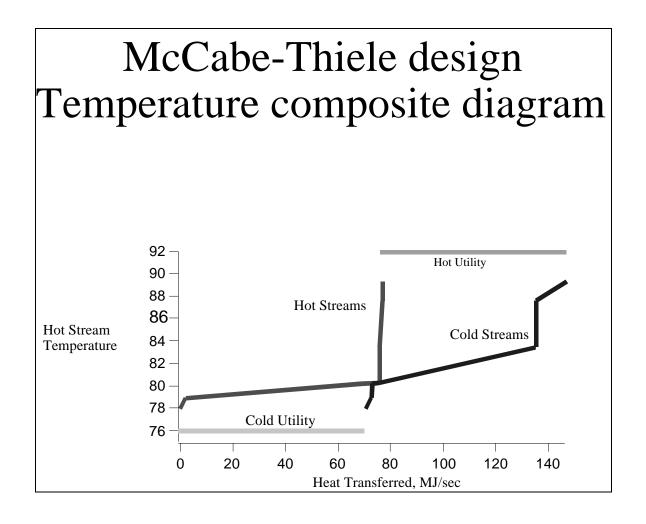
Minimum Work of Separation, W_{min} Difference of Gibbs Free Energy

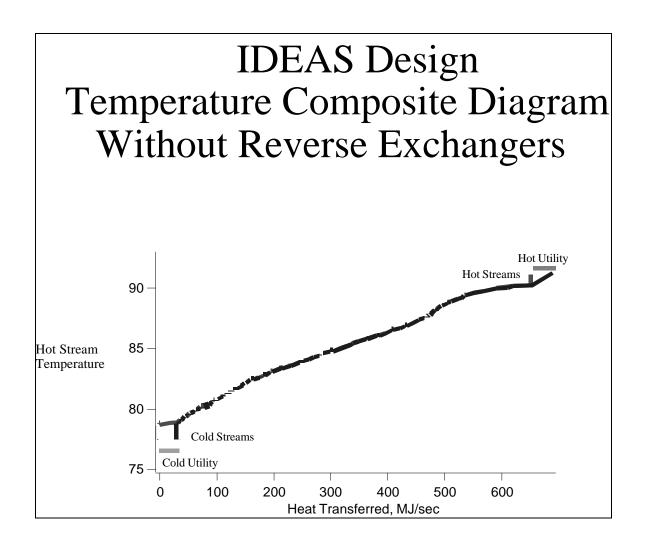
Process work, W_{proc}

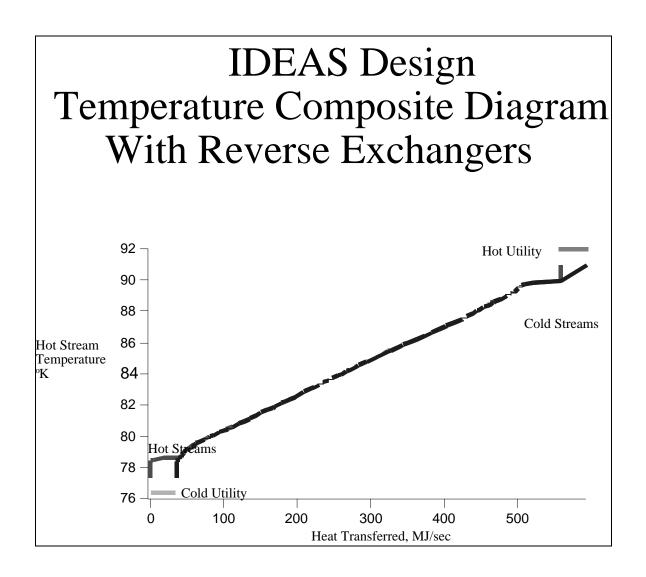
Work equavalent of heat (Carnot cycle)

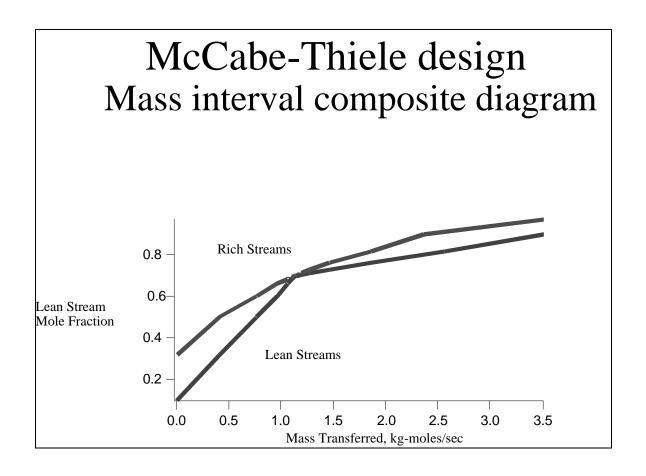
$$W_{proc} = Q_{proc} \frac{(T_{proc} - T_o)}{T_o}$$

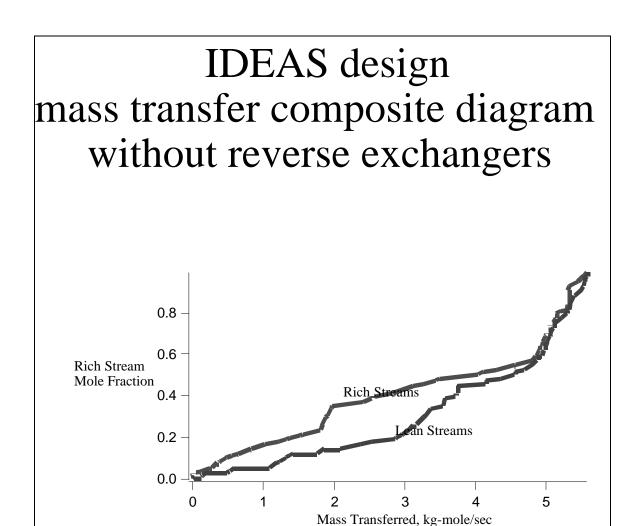
Results Case 1							
Utility Use (W _{proc})	McCabe- Thiele 12.24	IDEAS NR 6.14	IDEAS R 5.96				
O_d^{\ddagger}	0.15	0.29	0.30				
Cost (\$/sec)	79,800	39,700	38,800				



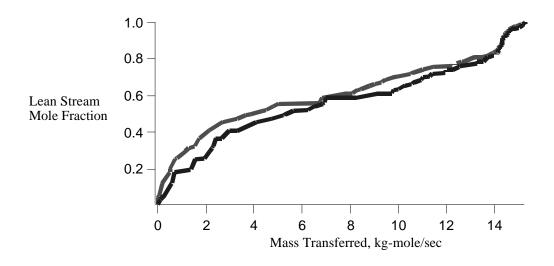




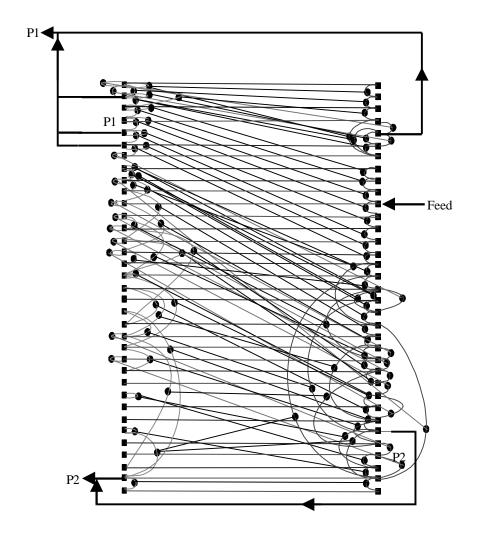




IDEAS design mass transfer composite diagram with reverse exchangers



IDEAS separation network



Case 2:

	McCabe- Thiele	IDEAS NR	IDEAS R
Hot Utility	11.74	9.11	8.76
Cold Utility	27.24	24.61	24.26
Cost (\$/sec)	29,300	26,300	25,900

Globally Optimal
Distillation Networks
Minimum Plate Area
for Fixed Utility Cost

Example:

Feed: 10 kg-moles/sec: 70 mole % N₂, 30% O₂, saturated vapor

Distillate: 7.5 kg-moles/sec: 90% N₂, saturated vapor

Bottoms: 2.5kg-moles/sec: 90% O₂ saturated liquid

Constant relative Volatility: 4.173

Pressure: 1.013x10⁵ Pa

Utility prices:

Hot utility: \$105/MJ at 92°K

Cold utility: \$1030/MJ at 76°K

Total utility cost: \$50,000/sec

Objective

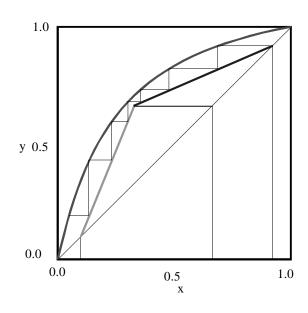
Determine the *globally* minimum plate area for a given separation and utility cost using IDEAS and compare with a McCabe-Thiele design

McCabe-Thiele design

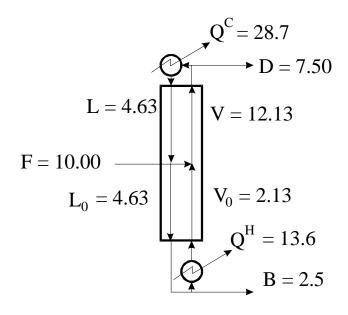
Determining heat loads

$$Fh_F + Q^H = Bh_B + Dh_D + Q^C$$

$$U = c_H Q^H + c_C Q^C$$



McCabe-Thiele design for fixed total utility cost



$$(L/V) = 1.33(L/V)_{min}$$

Results			
	McCabe- Thiele	IDEAS NR	IDEAS R
Plate Area	41.1	20.3	20.3
Total Utility Cost =	\$50,000/sec		

Globally Optimal Reaction/Distillation Networks

Motivation

- Reactor and Distillation networks impact waste generation at two levels: Processing and Recycling
- Design of such networks typically pursued through convex and/or mixed integer programs which do not guarantee global optimality.

Infinite DimEnsionAl State-space (IDEAS)

States: Composition, Enthalpy

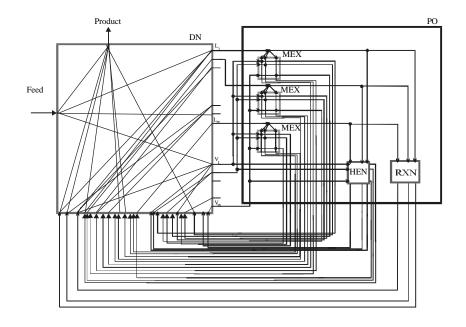
Infinite number of states

Distribution Network (DN): allows for stream mixing

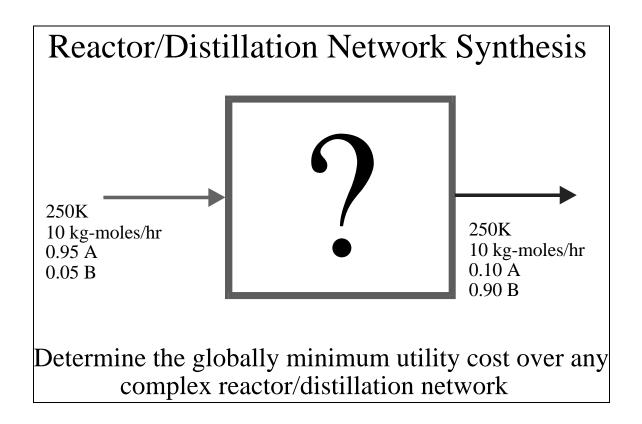
Process Blocks (RXN, MEN, HEN) external to DN

Includes ALL possible designs

Convex Programs: Local optima are global



IDEAS Representation of a Reaction/Distillation Network



Assumptions:

Process is isobaric at 5.875x10⁵ Pa

MEX's incorporate equilibrium plates

HEX's do not result in mixed phase streams

States are feeds, products, saturated liquids/vapors

Local mixing

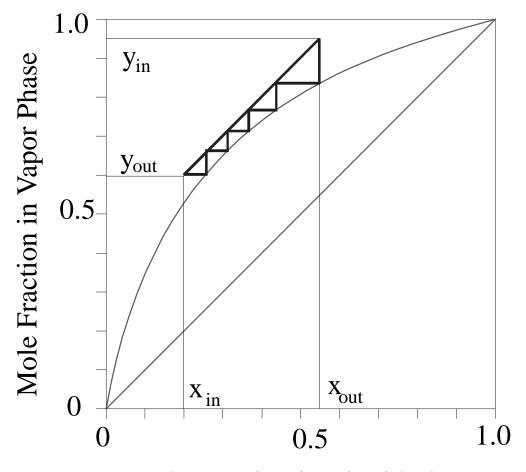
Reverse exchangers allowed

Constant relative volatility: 4.422; Chemical equilibrium constant: 1.7

Hot utility available at 92K for \$105/MJ, cold utility at 76K for \$1030/MJ

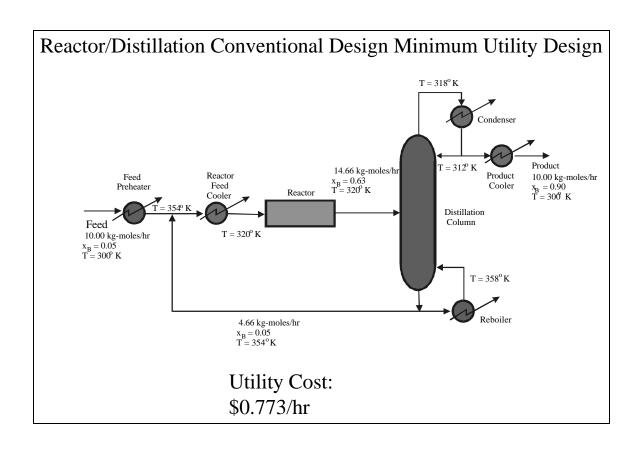
Reactors operate isothermally

Forward reaction is exothermal:) $H_r = -1.62-0.0027(T-298)$ MJ/kg-mole

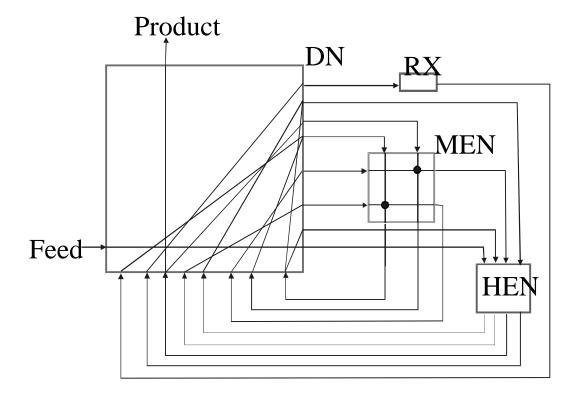


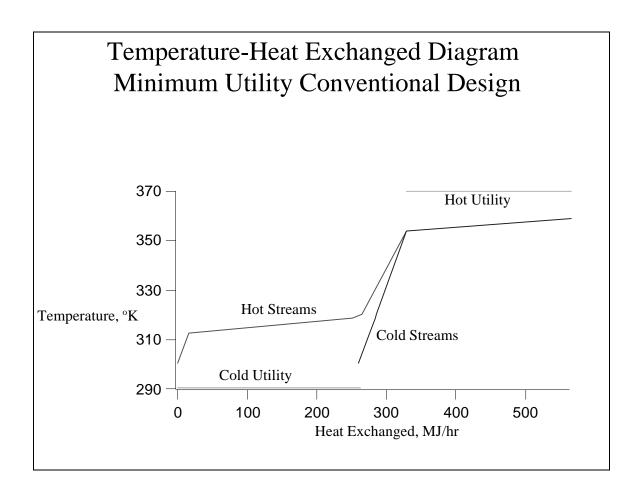
Mole Fraction in Liquid Phase

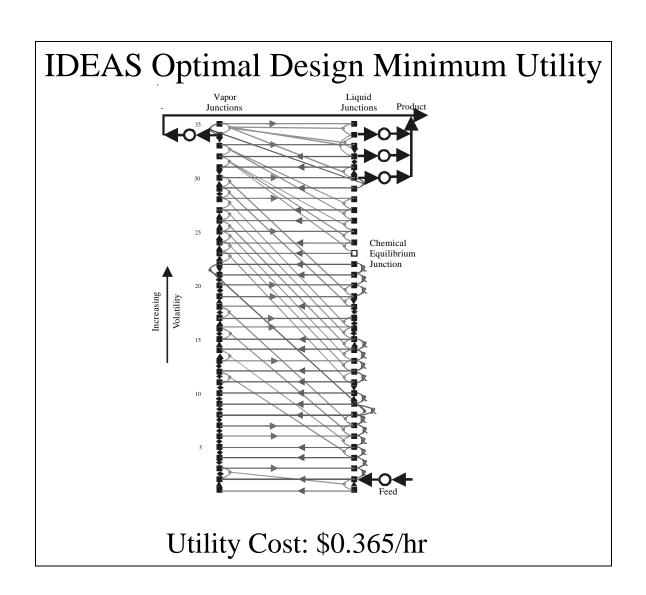
Reverse Exchangers

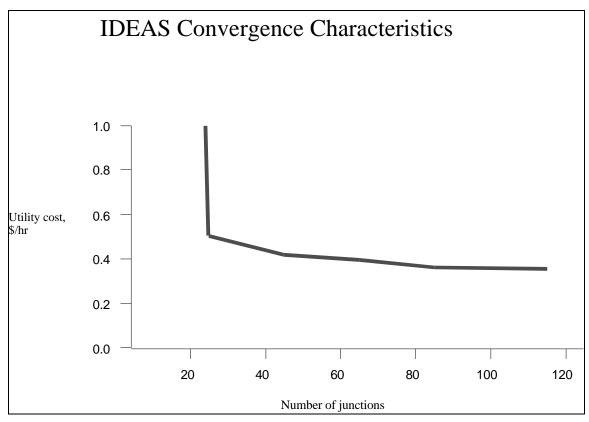


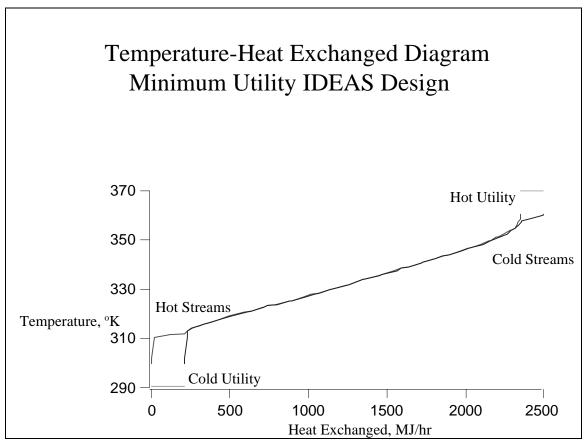
State-space representation of conventional optimal design

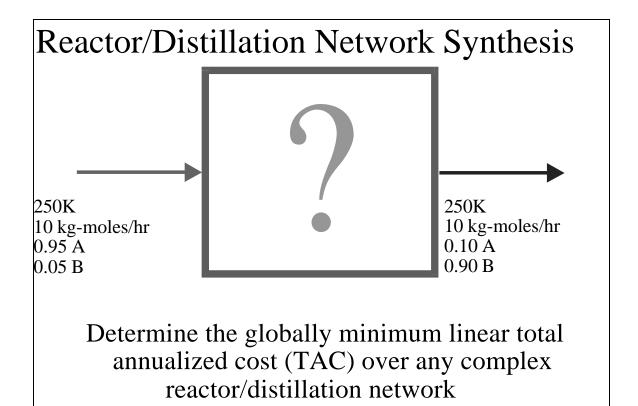












Additional Assumptions:

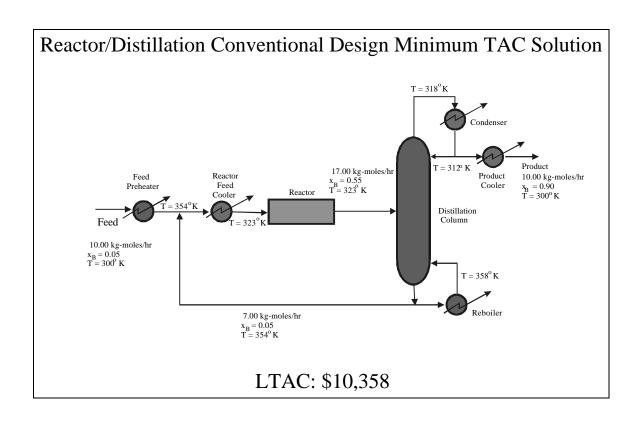
Capital cost of plates linearly proportional to plate area = $$50/m^2$

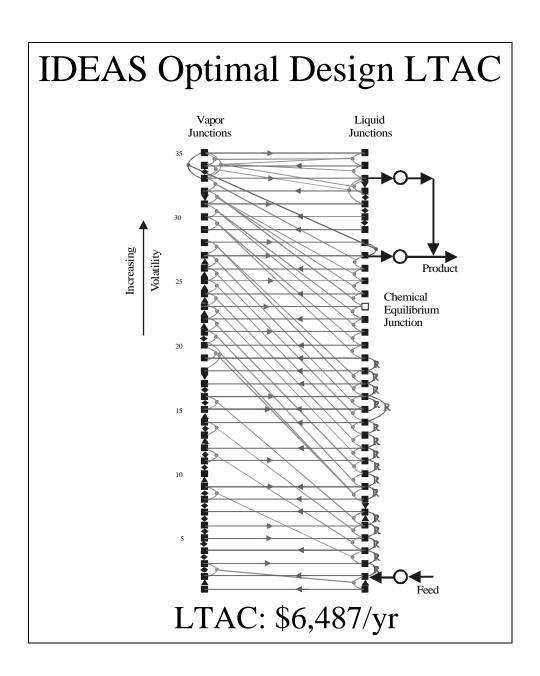
Capital cost of reactors linearly proportional to reactor volume = $$100/m^3$

TAC is given by

TAC = Annual utility cost + (Total capital cost)/ (Project life)

Project life = 7 years





Conclusions

Infinite DimEnsionAl State-space (IDEAS) process representation includes all possible processes

Resulting problem formulations are convex

IDEAS designs are flexible and may be used to represent a wide variety of processes

For reaction/distillation networks, IDEAS designs have lower utility cost and TAC than conventional designs

ACKNOWLEDGEMENTS

Mahmoud El-Halwagi Ashish Gupta Stevan Wilson James Drake